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Key indicators

Single-crystal X-ray study
 $T = 296$ K
 Mean $\sigma(\text{C}-\text{C}) = 0.010$ Å
 Disorder in main residue
 R factor = 0.107
 wR factor = 0.313
 Data-to-parameter ratio = 15.5

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

(Z)-6-[[1,3-Dihydroxy-2-(hydroxymethyl)propan-2-yl-amino]methylene]-4-[(E)-(4-ethylphenyl)diazenyl]-cyclohexa-2,4-dienone

The title compound, $\text{C}_{19}\text{H}_{23}\text{N}_3\text{O}_4$, adopts the keto-amine tautomeric form, with the H atom located on N rather than on O. This H atom is involved in a strong intramolecular hydrogen bond. The configuration around the azo $\text{N}=\text{N}$ double bond is *trans* and the dihedral angle between the two aromatic rings is $8.94(2)^\circ$. The molecules, with intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonding, are linked by intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds, forming a three-dimensional network.

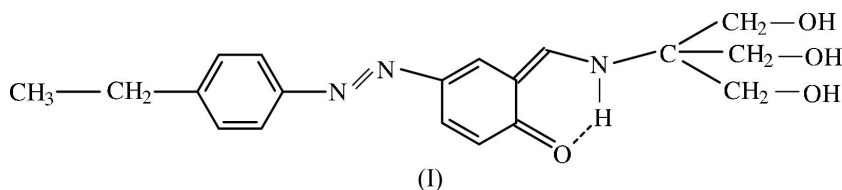
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Comment

Optically active azobenzene polymers are very important functional materials because of their photoresponsive properties. The pendant azobenzene groups in these materials behave as both photoresponsive chromophores and mesogens (Labarthe *et al.*, 1999). Recently, the formation of holographic gratings with polymeric azobenzene liquid crystals (LC) containing only azobenzene groups, where each group is mesogenic and photoresponsive, was reported (Yamamoto *et al.*, 2001). We present here the crystal structure of the title compound, (I).



In the azo group of (I), the $\text{N}1-\text{C}6$ and $\text{N}2-\text{C}9$ bond lengths [$1.428(9)$ and $1.428(8)$ Å, respectively] indicate significant single-bond character, whereas the $-\text{N}=\text{N}-$ bond length [$1.253(7)$ Å] is indicative of significant double-bond character. Similar values have been observed in other *trans*-azo compounds (Odabaşoğlu *et al.*, 2003; Koşar *et al.*, 2004). The 4-ethyl group shows disorder and was studied in two different orientations as groups *Ca* and *Cb* (*Ca* is the $\text{C}1a-\text{C}2a$ group and *Cb* is the $\text{C}1b-\text{C}2b$ group), with occupancy factors of 0.396 (14) and 0.604 (14), respectively (Fig. 1). In the molecule of (I), the $\text{C}12=\text{O}1$ and $\text{C}11-\text{C}15$ bond lengths are $1.292(7)$ and $1.416(8)$ Å, respectively (Table 1), and these are in agreement with the corresponding distances in 4-[(4-methylphenyl)diazenyl]-6-methoxy-2-[[tris(hydroxymethyl)methyl]aminomethylene]-3,5-dien-1(2*H*)-one [$1.279(5)$ and $1.406(6)$ Å, respectively; Koşar *et al.*, 2004].

In the molecule of (I), the dihedral angle, θ_1 , between the mean planes of the $\text{C}3-\text{C}8$ benzene ring and the $\text{C}6-\text{N}1=\text{N}2-\text{C}9$ azo bridge is $12.14(3)^\circ$, and the angle, θ_2 , between the planes of the $\text{C}6-\text{N}1=\text{N}2-\text{C}9$ azo group and

the substituted C9–C14 phenyl ring is 8.55 (2)°. The angle, θ_3 , between the planes of the two rings is 8.94 (2)°.

In the crystal structure, compound (I) has strong intramolecular N–H···O [N3···O1 = 2.652 (6) and N3···O2 = 2.712 (7) Å] and weak intermolecular O–H···O [O3···O1 = 2.685 (6), O2···O3 = 2.718 (6) and O4···O1 = 2.726 (6) Å] hydrogen bonds (see Table 2 for details).

Experimental

Compound (I) was prepared as described in the literature (Odabaşoğlu *et al.*, 2003) using salicylaldehyde, 4-ethylaniline and tris(hydroxymethyl)aminomethane as starting materials. The product was recrystallized from ethanol and well shaped crystals were obtained by slow evaporation of the same solution (yield 85%; m.p. 463–465 K).

Crystal data

C₁₉H₂₃N₃O₄
M_r = 357.40
 Monoclinic, *P*2₁/*c*
a = 18.441 (4) Å
b = 7.8992 (10) Å
c = 13.014 (3) Å
 β = 95.516 (16)°
V = 1887.0 (6) Å³
Z = 4

D_x = 1.258 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 7373 reflections
 θ = 1.6–23.5°
 μ = 0.09 mm⁻¹
T = 296 K
 Plate, brown
 0.50 × 0.40 × 0.12 mm

Data collection

Stoe IPDS 2 diffractometer
 ω scans
 Absorption correction: none
 9340 measured reflections
 3690 independent reflections
 1336 reflections with *I* > 2σ(*I*)

*R*_{int} = 0.129
 θ_{\max} = 26.0°
h = -22 → 22
k = -9 → 9
l = -13 → 16

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.107
wR (*F*²) = 0.313
S = 0.93
 3690 reflections
 238 parameters

H atoms treated by a mixture of independent and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.157P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.47 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.31 \text{ e } \text{Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

| | | | |
|-------------|------------|--------|-----------|
| C6–N1 | 1.428 (9) | C15–N3 | 1.295 (7) |
| C9–N2 | 1.428 (8) | C16–N3 | 1.482 (7) |
| C11–C15 | 1.416 (8) | N1–N2 | 1.253 (7) |
| C12–O1 | 1.292 (7) | | |
| C6–N1–N2–C9 | –175.9 (5) | | |

Table 2

Hydrogen-bond geometry (Å, °).

| <i>D</i> –H··· <i>A</i> | <i>D</i> –H | H··· <i>A</i> | <i>D</i> ··· <i>A</i> | <i>D</i> –H··· <i>A</i> |
|--------------------------|-------------|---------------|-----------------------|-------------------------|
| O2–H2···O3 ⁱ | 0.83 (2) | 1.93 (4) | 2.718 (6) | 158 (8) |
| O3–H3···O1 ⁱⁱ | 0.84 (2) | 1.85 (2) | 2.685 (6) | 177 (8) |
| N3–H3A···O1 | 0.93 (8) | 2.00 (7) | 2.652 (6) | 125 (6) |
| N3–H3A···O2 | 0.93 (8) | 2.20 (7) | 2.712 (7) | 114 (5) |
| O4–H4···O1 ⁱⁱ | 0.82 (2) | 1.96 (3) | 2.762 (6) | 163 (8) |

Symmetry codes: (i) $-x + 2, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x, -y - \frac{1}{2}, z + \frac{1}{2}$.

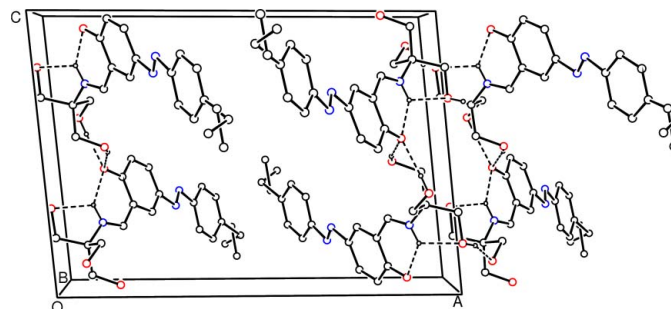


Figure 2

A packing diagram for (I), showing the hydrogen bonds (dashed lines). Both disorder components are shown. H atoms not involved in hydrogen bonding have been omitted for clarity.

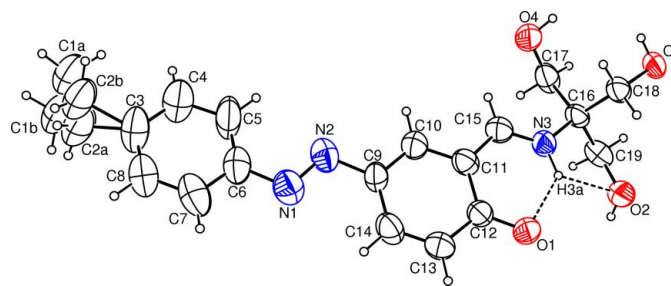


Figure 1

A view of the molecule of (I), with the atom-numbering scheme and 50% probability displacement ellipsoids. Dashed lines indicate the intramolecular hydrogen bonds. Both disorder components are shown.

The crystal of (I) used for data collection was not of good quality and was twinned. Accordingly, the quality of the crystallographic data was not as desired and this is the reason for the low quality of some crystallographic parameters. All attempts to find a better crystal failed. All H atoms bound to C atoms were refined using a riding model, with C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic C atoms, C–H = 0.97 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for methylene C atoms, and C–H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl C atoms. The H atoms of the hydroxyl O atoms were refined with O–H restrained to 0.83 (2) Å and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$. The H atom bonded to N was refined freely, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* for Windows (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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References

- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Koşar, B., Albayrak, Ç., Odabaşoğlu, M. & Büyükgüngör, O. (2004). *Acta Cryst. E* **60**, o246–o247.
 Labarthe, F. L., Rochon, P. & Natansohn, A. (1999). *Appl. Phys. Lett.* **75**, 1377–1379.

Odabaşođlu, M., Albayrak, Ç., Büyükgüngör, O. & Goesmann, H. (2003). *Acta Cryst. C* **59**, o234–o236.

Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

Stoe & Cie (2002). *X-AREA* (Version 1.18) and *X-RED32* (Version 1.04). Stoe & Cie, Darmstadt, Germany.

Yamamoto, T., Ohashi, A., Yoneyama, S., Hasegawa, M., Tsutsumi, O., Kanazawa, A., Shiono, T. & Ikeda, T. (2001). *J. Phys. Chem. B*, **105**, 2308–2313.